

# High-affinity $\text{Ca}^{2+}$ -binding site inhibiting $\text{Ca}^{2+}$ release from sarcoplasmic reticulum

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Received 2 September 1988; revised version received 23 October 1988

$\text{Ca}^{2+}$  release from sarcoplasmic reticulum membranes, activated by alkaline pH occurs only when EGTA is present in the release medium. Addition of very low concentrations of  $\text{Ca}^{2+}$  to the medium inhibits  $\text{Ca}^{2+}$  release. The concentration of free  $\text{Ca}^{2+}$  required for 50% inhibition ranges from between 5 and 20 nM in different experiments and/or membrane preparations, irrespective of whether the free  $\text{Ca}^{2+}$  concentration is controlled by EGTA or CDTA. Other divalent cations such as  $\text{Mn}^{2+}$ ,  $\text{Ba}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Cd}^{2+}$  and  $\text{Mg}^{2+}$  also exert an inhibitory effect on  $\text{Ca}^{2+}$  release, with higher or lower potency than that of  $\text{Ca}^{2+}$ . The inactivation of  $\text{Ca}^{2+}$  release by  $\text{Ca}^{2+}$  is reversible. We suggest the involvement of high-affinity  $\text{Ca}^{2+}$ -binding sites in the control of  $\text{Ca}^{2+}$  release.

Sarcoplasmic reticulum;  $\text{Ca}^{2+}$  release

## 1. INTRODUCTION

$\text{Ca}^{2+}$  release from the sarcoplasmic reticulum (SR) plays an important role in excitation-contraction coupling [1]. Using skinned muscle fibers and isolated SR membranes, various treatments and stimuli, such as addition of  $\text{Ca}^{2+}$  [2], depolarization of the membrane [3], alteration in pH [4] or surface charge [5], chemical modifications of the membranes [6,7], or the presence of inositol trisphosphate [8], have been shown to induce  $\text{Ca}^{2+}$  release.

Recent studies indicate that there is more than one  $\text{Ca}^{2+}$ -release system in the sarcoplasmic reticulum, involving two or more types of divalent cation channels [6,7,9–16]. These channels differ

in: their divalent cation conductivity and specificity, their distribution in the different structural regions of the SR membrane, and also in their activation mechanism. One of these channels located in the SR-T-system junction, is the 'ligand-gated' channel which recently has been identified and purified [11,17,18]. This channel has a high conductance, and is activated by adenine nucleotides, ryanodine and  $\text{Ca}^{2+}$  and inhibited by  $\text{Mg}^{2+}$  and ruthenium red [11,14,15]. The presence of a second channel with a low conductance and in all SR structural regions has been demonstrated recently [15,16]. However, the control mechanism of this channel is not yet known.

We have studied  $\text{Ca}^{2+}$  release in unfractionated native SR membranes and have developed several different conditions for activation of  $\text{Ca}^{2+}$  release [4–7,19]. This  $\text{Ca}^{2+}$ -release system is present in all structural regions of the SR membranes, and apparently operates via a channel different from the junctional  $\text{Ca}^{2+}$ -release channel. Our results suggest that the opening of this non-junctional channel is controlled by a positively charged, internally located, amino group(s).

It is not clear, however, from the different mechanisms proposed for the physiological release

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*Abbreviations:* DCCD, *N,N'*-dicyclohexyl carbodiimide; Tricine, *N*-[2-hydroxy-1,1-bis-(hydroxy-methyl)ethyl]glycine; Mops, 3-(*N*-morpholino)propanesulfonic acid; CDTA, *trans*-1,2-diamino-cyclohexane-*N,N,N',N'*-tetraacetic acid; RR, ruthenium red; TPB<sup>−</sup>, tetraphenylboron; SR, sarcoplasmic reticulum

of  $\text{Ca}^{2+}$  what is the control mechanism for channel closing. The inactivation of the channel should be different from the usual voltage- and time-dependent inactivation of excitable membranes.

In a previous study [4] we suggested that  $\text{Ca}^{2+}$  itself controls the closing of  $\text{Ca}^{2+}$ -release channel(s) in skeletal SR membranes. The control by  $\text{Ca}^{2+}$  of activation and inactivation of cardiac SR  $\text{Ca}^{2+}$ -release channel has been suggested by Fabiato [20]. In this communication we report on the involvement of high-affinity  $\text{Ca}^{2+}$ -binding sites in the inactivation of  $\text{Ca}^{2+}$  release from SR vesicles.

## 2. MATERIALS AND METHODS

### 2.1. Materials

ATP, EGTA, Tris, Mops, DCCD and CDTA were obtained from Sigma.  $^{45}\text{CaCl}_2$  was from Radiochemical Center, Amersham, and ruthenium red (97% pure) was from Fluka.

### 2.2. Methods

#### 2.2.1. Membrane preparations

Sarcoplasmic reticulum vesicles were prepared by the procedure of MacLennan [21], suspended to a final concentration of 20 mg protein/ml in a buffer containing 0.2 M sucrose, 10 mM Tricine, pH 8.0, and 1 mM histidine and stored at  $-70^\circ\text{C}$ . The protein concentration was determined by the Lowry et al. method [22].

#### 2.2.2. $\text{Ca}^{2+}$ efflux

$\text{Ca}^{2+}$  efflux from actively loaded SR vesicles was carried out as described in fig.1. The vesicles were loaded with  $\text{Ca}^{2+}$  for 15 min and then collected by centrifugation at  $100000 \times g$  for 20 min, and resuspended in 20 mM Mops, pH 6.8, and 100 mM KCl at a protein concentration of about 3 mg/ml. Since  $\text{TPB}^-$  precipitates  $\text{K}^+$ , all chemicals in the experiments with  $\text{TPB}^-$  were used as  $\text{Na}^+$  salts.

#### 2.2.3. Calculations of free concentrations of cations

Free concentrations were calculated with a computer program using the following apparent binding constants: at pH 6.8,  $\text{Ca-EGTA}$   $1.94 \times 10^6$ ;  $\text{Ca-CDTA}$   $3.26 \times 10^7$ ;  $\text{Mg-EGTA}$   $23.8$ ; and at pH 8.0,  $\text{Ca-EGTA}$   $4.28 \times 10^8$ ;  $\text{Ca-CDTA}$   $6.24 \times 10^8$ ;  $\text{Mg-EGTA}$   $8.8 \times 10^2$ ;  $\text{Mn-EGTA}$   $5.4 \times 10^9$ ;  $\text{Cd-EGTA}$   $7.48 \times 10^{15}$ ;  $\text{Cu-EGTA}$   $2.1 \times 10^{15}$  [23].

## 3. RESULTS

Dilution of  $\text{Ca}^{2+}$ -loaded SR vesicles into a basic medium containing EGTA, activates  $\text{Ca}^{2+}$  release [4]. The amount of  $\text{Ca}^{2+}$  released is a function of the EGTA concentration in the release medium (fig.1). Under the conditions used in this experiment (pH 8.0 and 0.1 mg SR/ml), maximal  $\text{Ca}^{2+}$

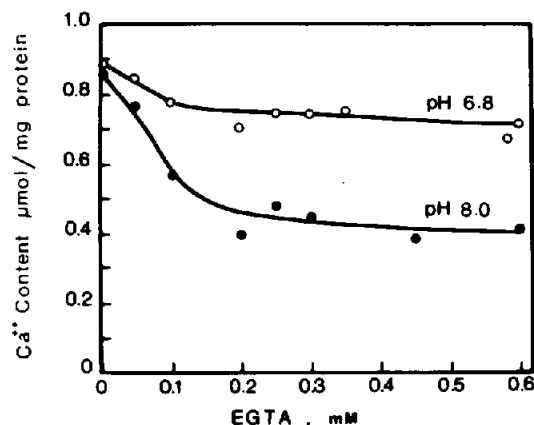


Fig.1. pH-induced  $\text{Ca}^{2+}$  efflux as a function of EGTA concentration in the release medium. SR vesicles were actively loaded with  $^{45}\text{CaCl}_2$  as in [24] and collected by centrifugation and resuspended as described in section 2.  $\text{Ca}^{2+}$  efflux was assayed by 50-fold dilution of the membranes into 100 mM NaCl and 20 mM Mops, pH 6.8 (○) or 20 mM Tris, pH 8.0 (●), in the absence or presence of the indicated concentration of EGTA.  $\text{Ca}^{2+}$  content was determined 30 s after dilution, by filtering 0.18 ml samples through  $0.3 \mu\text{m}$  nitrocellulose filters and washing with 5 ml of 0.15 M of KCl. Radioactivity on the filters was measured in a liquid scintillation counter.

release was obtained at 0.15 mM EGTA. In the absence of EGTA the amount of  $\text{Ca}^{2+}$  released at pH 8.0 was similar to that released in control vesicles incubated at pH 6.8.

The  $\text{Ca}^{2+}$  release, stimulated by alkaline pH, is also a function of the SR concentration in the release medium. About 85% and 7% of trapped  $\text{Ca}^{2+}$  was released upon 200- and 25-fold dilution, respectively (not shown). These results suggest that external  $\text{Ca}^{2+}$  is involved in the inhibition of  $\text{Ca}^{2+}$  release. Fig.2 shows that addition of  $\text{Ca}^{2+}$  at very low concentrations, to the release medium inhibits  $\text{Ca}^{2+}$  release. 50% inhibition was obtained at less than 10 nM in some membrane preparations and about 20 nM in others. Assuming that the total EGTA concentration is higher or has about 5% uncertainty, the concentration of free  $\text{Ca}^{2+}$  required for 50% inhibition ranges between 5 and 15 nM. Similar inhibition was obtained whether the free  $\text{Ca}^{2+}$  concentration was controlled by the chelator EGTA or CDTA, indicating that it was  $\text{Ca}^{2+}$  and not the  $\text{Ca}^{2+}$ -chelator complex that was responsible for the inhibition of  $\text{Ca}^{2+}$  release. The fact that  $\text{Ca}^{2+}$  release went from uninhibited to almost completely inhibited upon doubling the

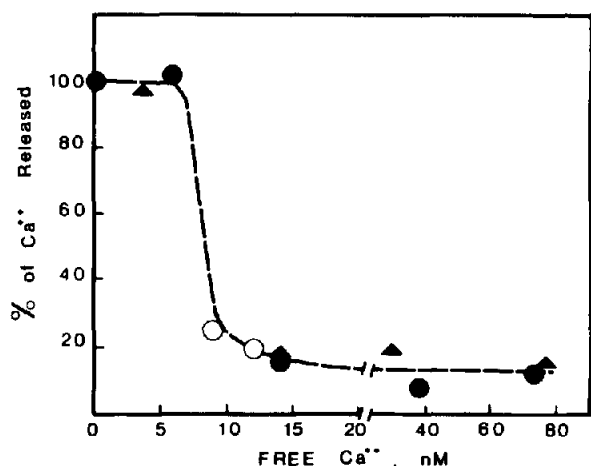


Fig.2.  $\text{Ca}^{2+}$  inhibition of  $\text{Ca}^{2+}$  release stimulated by elevated pH. SR vesicles were actively loaded with  $^{45}\text{CaCl}_2$  and assayed for  $\text{Ca}^{2+}$  efflux as in table 1 except that free  $\text{Ca}^{2+}$  at the indicated concentration was present in the release medium and the pH in each case was adjusted to 6.8 or 8.0.  $\text{Ca}^{2+}$  content was determined 2 min after dilution. Control activity (100%) was  $1.4 \mu\text{mol Ca}^{2+}/\text{mg protein}$ . Free  $\text{Ca}^{2+}$  concentrations were calculated as described in section 2. (○) and (●) indicate two different experiments in which free  $\text{Ca}^{2+}$  concentration was controlled by EGTA, while in (▲) by CDTA.

$\text{Ca}^{2+}$  concentration in the release medium, suggests positive cooperativity in the  $\text{Ca}^{2+}$  effect and, therefore, on the involvement of two or more  $\text{Ca}^{2+}$ -binding sites. The results suggest the involvement of high-affinity  $\text{Ca}^{2+}$ -binding sites in the control of  $\text{Ca}^{2+}$  release.

$\text{Ca}^{2+}$  also inhibits  $\text{Ca}^{2+}$  release induced by other means such as by TPB<sup>-</sup> [19] or by chemical modification of SR vesicles with acetic anhydride [6]. High  $\text{Ca}^{2+}$  concentrations (mM) were required for the inhibition of TPB<sup>-</sup>-induced  $\text{Ca}^{2+}$  release (not shown). This may be due to the TPB<sup>-</sup>-negative charge incorporated into the membranes [5]. Inhibition of  $\text{Ca}^{2+}$  released induced by acetic anhydride modification was obtained with micromolar concentrations of  $\text{Ca}^{2+}$  (not shown).

Table 1 shows that other cations besides  $\text{Ca}^{2+}$  exert an inhibitory effect on  $\text{Ca}^{2+}$  release.  $\text{Mn}^{2+}$  and  $\text{Ba}^{2+}$  were inhibitory with similar potency to  $\text{Ca}^{2+}$ . On the other hand,  $\text{Cu}^{2+}$  and  $\text{Cd}^{2+}$  were 4–5 orders of magnitude more potent and  $\text{Mg}^{2+}$  was with about 4 orders of magnitude less effective. However, it is possible that the effect of  $\text{Cu}^{2+}$  and  $\text{Cd}^{2+}$  is due to the release of complexed endogenous  $\text{Ca}^{2+}$  from EGTA.

Table 1

Effect of different divalent cations on  $\text{Ca}^{2+}$  release

Additions to $\text{Ca}^{2+}$ release medium	Free cation concentration, M	$\text{Ca}^{2+}$ released % of control
None	—	51
$\text{Ca}^{2+}$ , 0.95 mM	$4.4 \times 10^{-8}$	10
$\text{Mg}^{2+}$ , 5 mM	$1.4 \times 10^{-3}$	10
$\text{Ba}^{2+}$ , 0.95 mM	—	39
$\text{Ba}^{2+}$ , 1.2 mM	—	7
$\text{Mn}^{2+}$ , 0.95 mM	$3.4 \times 10^{-9}$	10
$\text{Mn}^{2+}$ , 1.2 mM	$2.0 \times 10^{-4}$	3
$\text{Cd}^{2+}$ , 0.95 mM	$7.5 \times 10^{-15}$	2
$\text{Cu}^{2+}$ , 0.8 mM	$1.9 \times 10^{-15}$	36
$\text{Cu}^{2+}$ , 0.95 mM	$8.8 \times 10^{-15}$	10

SR vesicles were actively loaded with  $\text{Ca}^{2+}$  and assayed for  $\text{Ca}^{2+}$  efflux as in fig.1, except that EGTA concentration was 1 mM and the indicated divalent cations were present in the release medium. Free cation concentrations were calculated as described in section 2.  $\text{Ca}^{2+}$  content was determined 2 min after dilution.  $\text{Ca}^{2+}$  content (100%) was between 3.2 and  $3.5 \mu\text{mol}/\text{mg protein}$  for all samples except for the  $\text{Mn}^{2+}$  containing samples which was 3.7 and  $4.2 \mu\text{mol}/\text{mg protein}$  for free concentrations of 3.4 nM and 200  $\mu\text{M}$ , respectively

Table 2 shows that the polycationic dye, ruthenium red, is also an inhibitor of  $\text{Ca}^{2+}$  release through the non-junctional channel activated by alkaline pH. Ruthenium red has been shown to inhibit the  $\text{Ca}^{2+}$ -induced  $\text{Ca}^{2+}$  release that operates through the ligand gated-junctional  $\text{Ca}^{2+}$ -release channel [11].

The inactivation of  $\text{Ca}^{2+}$  release by  $\text{Ca}^{2+}$  is reversible. Fig.3 shows the time course of reactivation of  $\text{Ca}^{2+}$ -inhibited  $\text{Ca}^{2+}$  release, produced by the addition of EGTA to a  $\text{Ca}^{2+}$ -containing

Table 2

Ruthenium red inhibition of pH-induced  $\text{Ca}^{2+}$  release

Ruthenium red ( $\mu\text{M}$ )	$\text{Ca}^{2+}$ released/min % of $\text{Ca}^{2+}$ control
0	46
2.5	35
5	16
15	8

SR vesicles were loaded with  $\text{Ca}^{2+}$  and assayed for  $\text{Ca}^{2+}$  efflux as in table 1 except that ruthenium red at the indicated concentration was present in the SR dilution medium.  $\text{Ca}^{2+}$  content was determined 1 min after dilution. Control activity (100%) was  $1.3 \mu\text{mol Ca}^{2+}/\text{mg protein}$

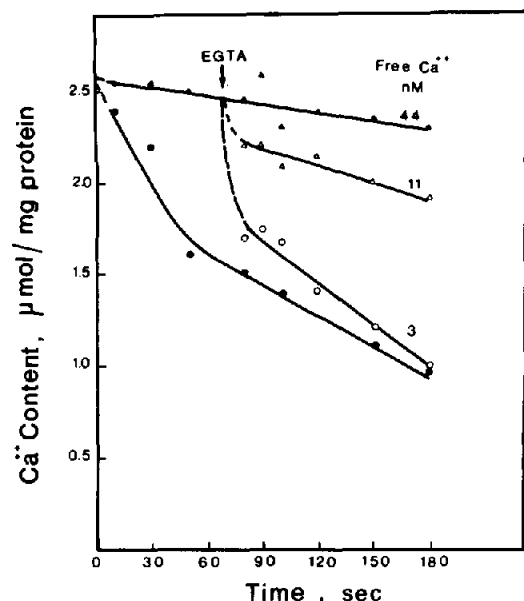


Fig.3.  $\text{Ca}^{2+}$  inhibition of pH-induced  $\text{Ca}^{2+}$  release can be reversed by EGTA. SR vesicles were actively loaded with  $^{45}\text{CaCl}_2$  and assayed for  $\text{Ca}^{2+}$  efflux as in table 1 except that the loaded vesicles were diluted into release medium containing 1 mM EGTA (●) or 1 mM EGTA and 0.95 mM  $\text{CaCl}_2$  (▲) (free  $[\text{Ca}^{2+}]$  is 44 nM). After 70 s EGTA (0.148 or 0.714 mM) was added to the  $\text{Ca}^{2+}$ -containing sample to lower the free  $\text{Ca}^{2+}$  concentration to 11 nM (Δ) or 3 nM (○), respectively.

medium.  $\text{Ca}^{2+}$ -loaded vesicles, diluted into a medium containing 44 nM free  $\text{Ca}^{2+}$ , retained most of their internal  $\text{Ca}^{2+}$ . However, when the free  $[\text{Ca}^{2+}]$  was decreased to 11 nM or 3 nM,  $\text{Ca}^{2+}$  is rapidly released. Thus, the results indicate that the transition from an inactive to an active  $\text{Ca}^{2+}$ -release system is very rapid.

Inactivation of active  $\text{Ca}^{2+}$  release was also obtained by the addition of  $\text{Ca}^{2+}$  to an EGTA-containing release medium. Inactivation of  $\text{Ca}^{2+}$  release could be halted immediately at any time during its time course by the addition of  $\text{Ca}^{2+}$  (not shown). The rate constants for activation and inactivation of  $\text{Ca}^{2+}$  release could not be determined due to experimental limitations in resolving the kinetics of these two processes.

#### 4. DISCUSSION

The study presented here suggests the involve-

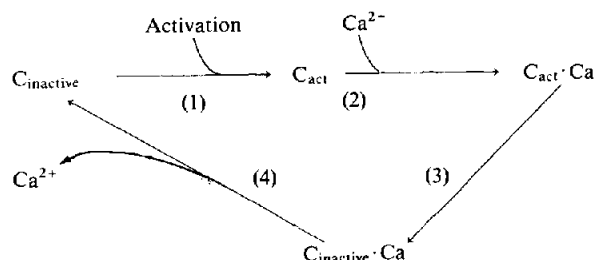
ment of high-affinity  $\text{Ca}^{2+}$ -binding sites in the control of  $\text{Ca}^{2+}$  release. Less than 20 nM of  $\text{Ca}^{2+}$  are required to inhibit  $\text{Ca}^{2+}$  release activated by alkaline pH. However, this concentration is dependent on the SR concentration in the release media.

The source of the inactivating  $\text{Ca}^{2+}$  could be intra- or extravesicular. Even in the presence of low SR concentrations and 50  $\mu\text{M}$  EGTA no  $\text{Ca}^{2+}$  release was observed (fig.1), suggesting that inhibition can be caused by the small amount of  $\text{Ca}^{2+}$  released upon activation. On the other hand, addition of  $\text{Ca}^{2+}$  to the external medium also prevented the release of  $\text{Ca}^{2+}$  (fig.2). The inhibitory effect of  $\text{Ca}^{2+}$  is reversible (fig.3).

Our previous studies [5–7,19] suggest that  $\text{Ca}^{2+}$  release, activated by various treatments, probably operate through one or more types of  $\text{Ca}^{2+}$ -release channels, present in SR membranes [12–16]. We suggest, therefore, that  $\text{Ca}^{2+}$  by its binding to specific sites led to inactivation of  $\text{Ca}^{2+}$ -release channel(s), and that the non-junctional channel is much more sensitive to  $\text{Ca}^{2+}$ .

The  $\text{Ca}^{2+}$ -binding sites are probably located within a protein because internal  $\text{Ca}^{2+}$  did not prevent the release of  $\text{Ca}^{2+}$  upon activation and this  $\text{Ca}^{2+}$  release is inhibited by DCCD but not with the water-soluble carboxylic group reagents [24]. We propose, therefore, that a carboxylic group (or groups) exposed on activation of  $\text{Ca}^{2+}$  release is involved in the inhibition of  $\text{Ca}^{2+}$  release.

The  $\text{Ca}^{2+}$ -binding sites, controlling the inactivation of the  $\text{Ca}^{2+}$  channel(s), have a very high affinity ( $K_d < 10$  nM). However, this high affinity for  $\text{Ca}^{2+}$  might be expected to lead to channel closing immediately following the first burst of  $\text{Ca}^{2+}$  released. This, however, would not be the case if there is a rate limiting step such as the slow binding of  $\text{Ca}^{2+}$  (reaction (2) in the model of scheme 1) or if the complex channel·Ca must undergo a slow conformational change(s) (reaction (3) in the model) in order to reach its closed state. We tried to examine these two possibilities by following: (i) the rate of  $\text{Ca}^{2+}$  inhibition of preactivated  $\text{Ca}^{2+}$  release and (ii) the rate of activation of  $\text{Ca}^{2+}$ -inactivated  $\text{Ca}^{2+}$  release, produced by the addition of EGTA (fig.3). In both experiments the rate was too fast to follow by the filtration technique. However, this does not rule out the involvement of such rate limiting step(s) in the  $\text{Ca}^{2+}$ -dependent inactivation.



Scheme 1. for time- and  $\text{Ca}^{2+}$ -dependent inactivation of  $\text{Ca}^{2+}$  release.  $C_{\text{inactive}}$  and  $C_{\text{act}}$  indicate inactivated and activated channel, respectively.

It should be mentioned that it is possible that the inactivation proceeds through a multi-state reaction. A similar mechanism for time- and  $\text{Ca}^{2+}$ -dependent inactivation of  $\text{Ca}^{2+}$ -induced  $\text{Ca}^{2+}$  release in cardiac SR has been suggested by Fabiato [20]. A  $\text{Ca}^{2+}$ -dependent inactivation of the voltage-gated  $\text{Ca}^{2+}$  channel has been demonstrated in a variety of excitable cells [25].

Studies of the  $\text{Ca}^{2+}$  binding and inactivation of  $\text{Ca}^{2+}$  release in SR membranes could provide further insight into the control mechanism for the closing of the  $\text{Ca}^{2+}$  release channel(s).

**Acknowledgements:** We thank Professor D.M. Chipman for his useful suggestion during the course of this work. This research was supported by the Fund for Basic Research Administered by The Israel Academy of Sciences and Humanities.

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